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Student answers will vary, but may include questions about 2. ocean formation, composition of ocean water, or water 3. temperature and pressure. Use your book or a dictionary to define atmosphere. Earth's air, which is made of a thin layer of gases, solids, and liquids; forms a protective layer around the planet and is divided

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*Glencoe Physical Science
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*For example, hydrocarbons
are not acids even though
their formulas contain H.
Hydrocarbons are easy to
recognize since their
formulas contain only C and
H. Acids, Bases, and Salts
In solution carbon dioxide,
CO₂, reacts with water to
form a small amount of
carbonic acid through the
equilibrium*

*Chem 1201 Exam 2 Flashcards
| Quizlet*

Q14: Expect the following

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salts to break down and give chemical equations: (3) A)

Sodium chloride. B)

magnesium sulfate. C)

potassium carbonate. Q15:

The acid name is as follows:

(3) A) cottage cheese. B)

bee sting. C) lemon juice.

Q16: The atomic numbers of

F, Na and Ne are 9, 10 and

11. Why do Na and F interact

so much with each other?

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