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10 Important

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Questions Short
Answer Type.

Question 1. (a) Give Arrhenius definition of an acid and a base. (CBSE 2006) (b)

Choose a strong acid and strong base from the following: CH_3COOH , NH_4OH , KOH , HCl Answer: (a) Arrhenius ...

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questions on the
difference between
acids, bases and salts,
indicators, litmus
paper test,
neutralisation and

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answers and
explanations to 11
multiple choice
questions, answers to
3 very short answer
questions ...

Question Bank for 7th
Class Science Acids,
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CBSE Class 10 Science
Book Chapter 2

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CBSE in the new
Exam pattern. At the
end of Multiple
Choice Questions, the
answer key has also
been provided for
your reference.

MCQ Questions for
Class 7 Science
Chapter 5 Acids,
Bases ...

a) Limewater is the
common name for a

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diluted solution of
calcium hydroxide
c) Range 1-3 d) Strong
Base is NaOH and
Weak Base is NH_4OH
e) A hydrated salt is
copper sulphate
crystals ($\text{CuSO}_4 \cdot 5\text{H}_2\text{O}$).
f) A universal
indicator is a mixed
indicator of organic
chemicals which not
only shows whether
the given solution is

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acidic or basic, but
also shows the
approximate pH
values by giving a
wide ...

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Chapter Wise
Important Questions

...

Short Answer type
Question 10 Give a
reason why (a) Great
care should be taken

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while handling
laboratory acids and
bases. (b) Solid
powder of baking
soda does not turn
dry litmus paper to
blue in color.

Question 11 Define
acid rain and how it is
produced? What are
its harmful effect?

Question 12 Define
neutralization
reaction with an

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example.
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Key This will be fine
as soon as knowing
the chapter 19 acids
bases salts answer
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Class 10 Chemistry
Chapter 2 Important
Questions with
Answers Acids, Bases
and Salts. Acids,
Bases and Salts Class
10 Important
Questions Very Short
Answer Type.

Question 1. How is
the concentration of
hydronium (H_3O^+)
ions affected when a
solution of an acid is

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diluted? (2017 OD)

Answer: When an
acid solution is
diluted with water
then ...

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Class 10 Important
Questions with ...

Access Answers of
Science NCERT class
10 Chapter 2 – Acids
Bases and Salts. In-
text questions set 1

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—> Page number 18.

1. You are given three
test tubes.

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...

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helpful in last minute
question and
Answers which has

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Acids ...

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Chapter 5 Acids,
Bases and Salts to
revise the concepts in
the syllabus
effectively and
improve your

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chances of securing high marks in your board exams. Acids, Bases and Salts Class 7 MCQs Questions with Answers.

Choose the correct answer: Question 1.

An oxide is acidic and has a pungent odour.

Acids and Bases
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Questions with
Answers to help
students understand
the concept very well.
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Class 7 MCQs

Questions with

Answers. Choose the
correct answer:

Question 1.

Chapter 19 Acids

Bases Salts Answer

Key

What are Acids,

Bases, and Salts? The

Swedish chemist

Svante Arrhenius

introduced the

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theory of ionization
and used this theory
to explain much
about the behavior of
acids and bases.

An Arrhenius acid is
defined as any
compound that
dissociates in
aqueous solution to
form _____ ions.

HNO

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Salts wkst-Key

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Hots Questions With
Answers. Question 1.

A student prepared solutions of (i) an acid and (ii) a base in two separate beakers. She forgot to label the solutions and litmus paper was not available in the laboratory. Since both the solutions

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were colourless, how
would she

distinguish between
the two? [NCERT
Exemplar] Answer:

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Questions ...

This consists of 1
mark Questions, 3
Mark Numericals
Questions, 5 Marks
Numerical Questions

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questions from Acids
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Bases and Salts

Chapter Wise
Important Questions
Class 10 Science
Short Answer Type
Questions[1] [2 Marks]

– Year 2015. 1.Name
the acid present in
the following: (i)
Tomato (ii) Vinegar
(iii ...

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MCQs for Class 10
Science Book Chapter
2 “ Acids, Bases and

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An acid is any substance whose aqueous solution is characterized by a sour taste, the ability to turn blue litmus red, and the ability to react with bases and certain metals to form salts. A Base ...

Answers about Acids
and Bases

Acids and bases

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interact with each other in what is called a neutralization reaction. The products of the reaction are a salt and water. The pH is "neutralized" to 7 if both the acid and base fully react.

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class 10 chapter 2
Acids, bases and salt

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Choice Questions
with Answers to
know their
preparation level.

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