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A solution of CaCl_2 was prepared by dissolving 0.0169g in 1 Kg of distilled water in (molar mass of $\text{Ca}^{2+} = 41\text{g mol}^{-1}$ - and $\text{Cl} = 35.5\text{g mol}^{-1}$ -). The molal depression constant of water is 2kg mol^{-1} -. The depression in freezing point of the solution is: Solution: Van't Hoff factor(i) of CaCl_2 is 3. $K_f = 2\text{ kg mol}^{-1}$ - Given, Mass of CaCl_2 (m ...

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Precipitation of heavy metal ions from the aq. solution of their salts : $\text{FeCl}_3 + 3\text{NH}_4\text{OH} \rightarrow \text{Fe}(\text{OH})_3 + 3\text{NH}_4\text{Cl}$ Brown ppt. $\text{AlCl}_3 + 3\text{NH}_4\text{OH} \rightarrow \text{Al}(\text{OH})_3 + 3\text{NH}_4\text{Cl}$ White ppt. $\text{CrCl}_3 + 3\text{NH}_4\text{OH} \rightarrow \text{Cr}(\text{OH})_3 + 3\text{NH}_4\text{Cl}$ Green ppt. Phosphorus: Allotropy of Phosphorus: a) White phosphorus: Translucent white waxy solid. Extremely reactive

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Frank Solutions for Class 9 Maths; ICSE Class. ICSE Class 6; ICSE Class 7; ICSE Class 8; ICSE Class 9; ICSE Class 10; ISC Class 11; ISC Class 12; IAS. ... 2nd Sep Shift-2 Maths; 9th Jan Shift-1 Chemistry; 9th Jan Shift-1 Physics; 9th Jan Shift-1 Maths; ... apart from Na^+ and Cl^- ions the solution of sodium chloride also contains H^+ and OH^- ...

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