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If we assume that both of them are first order, we get. $k_1 = \text{fractional yield of B} \times k_{av}$. $k_2 = \text{fractional yield of C} \times k_{av}$. If $k_1 > k_2$ then. A \rightarrow B main and. A \rightarrow C is side reaction. Let after a definite interval x mol/litre of B and y mol/litre of C are formed.

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