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2 (g) y2CO 2 (g) 7. Calculate the number of liters of oxygen gas needed to produce 15.0 liters of dinitrogen trioxide. Assume all gases are at the same conditions of temperature and pressure. 2N 2 (g) 1 3O 2 (g) y2N 2 O 3 (g) STOICHIOMETRY PRACTICE PROBLEMS 9 Name _____ Class _____ Date _____

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