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The resulting volume was 104 mL. Calculate the density, mole fraction, molarity, and molality of the solution. Assume water has a density of 1.00 g/cm³. 30. An aqueous antifreeze solution is 40.0% ethylene glycol (C₂H₆O₂) by mass. The density of the solution is 1.05 g/cm³. Calculate the ethylene ...

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Diisononyl phthalate | C₂₆H₄₂O₄ - PubChem
This chapter covers the pharmacology of chloroquine and hydroxychloroquine, which is similar for both drugs [1], but the details are different. For example, both drugs are partially excreted in feces, but the proportions differ slightly—8–10 % for chloroquine and 15–24 % for hydroxychloroquine.

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Gaseous Emissions - DieselNet
Forced degradation is a degradation of new drug substance and drug product at conditions more severe than accelerated conditions. It is required to demonstrate specificity of stability indicating methods and also provides an insight into degradation pathways and degradation products. elucidation of the structure of the degradation products.

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