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2 Answer.Key 2-1 Explore,
page 5 ... 2-1 Review and
Reinforcement, pages 8-9 .
1. energy 2. radiant 3.
chemical potential 4. joule

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5. created, destroyed 6.
natural gas ... 15.
chromatography 16.
electrolysis 17.
distillation 18. A magnet
could be used to separate
them

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. Reading the Periodic Table
. On the line at the left,
write the letter of the
appropriate location of each
group of elements on the

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periodic table below. Some letters will be used more than once. 1. carbon family 8. f-block elements 2. alkaline earth metals 9. noble gases 3. inner transition metals 10. p-block ...

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2) $4.75\text{g} \times 1 \text{ kg} = 0.00475$

kg. 1000g. 3) 0.000278 mole

$= 0.0582 \text{ m. } 0.00475 \text{ kg.}$ The

molality of a solution of

chlorine and water is

0.0362m. This solution

contains 3500g of water. How

much chlorine, in grams, was

used to prepare this

solution? ... 15-2 Review

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5-2 Review and Reinforcement
- MRS. Nairn

Name _____ Review and
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Solve each of the following problems as directed. Show all your work. 13. How many moles of magnesium are required to react with 2.0 mol of hydrochloric acid?

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10—2 Review and Reinforcement (continued) 9.
 $0.0621 \text{ mol SS} = (Q \$ \text{gK}2\text{S}$
10. $65.8\text{gMg} = \text{l. atoms Mg}$
Solve each of the following problems as directed. Show all your work. 11. Calculate the number Of atoms in 0.40 mol of sulfur- 12. Calculate the number of atoms in 2.30 mol of silver. G.oux 13. You

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have a tank of NO₂ gas.

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'16—1A Review and Reinforcement, pages 9-10 6. Answers Will Vary. When the rates Of forward and reverse reactions are equal, equilibrium is established. For the example of ... 15. The forward and reverse reactions are both occurring, but with no net change in concentration. -16—2 Review Reinforcement, pages 15-16

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the statement is true, write
"true." If it is false,
change the underlined word r
words to make it ... 15. 16.
20 Ionic compounds that
absorb water into their
solid structures fo When
naming an ionic substance,
the name of the ...

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